## Exponential Decay

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This is true because the exponential function is the only function that satisfies the differential equation $y^{\prime}=k y$.
As a result, remarkably little information is required to completely characterize the process.

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This quantity is called the half-life $t_{h}$ and is independent of the quantity present at the start:

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\frac{1}{2} y(0)=y(0) e^{k t_{h}} \quad \text { so } \quad \frac{1}{2}=e^{k t_{h}} \Rightarrow k=\frac{\ln \left(\frac{1}{2}\right)}{t_{h}}
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Then the quantity left at time $t$ is:

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y(t)=y(0) e^{k t}=y(0) \cdot e^{\ln \left(\frac{1}{2}\right)\left(\frac{t}{t_{h}}\right)}=y(0) \cdot\left(\frac{1}{2}\right)^{\left(\frac{t}{t_{h}}\right)}
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y(t)=y(0) \cdot\left(\frac{1}{2}\right)^{\left(\frac{t}{t_{h}}\right)}
$$

The quantity remaining after 180 days is:

$$
y(180)=300 \cdot\left(\frac{1}{2}\right)^{\left(\frac{180}{140}\right)}=123.05 \mathrm{mg}
$$

## Question 1

bismuth-210 has a half-life of 5 days. How much of a sample of 200 mg is left after 14 days?

\author{

1. 16.48 mg <br> 2. 92.57 mg <br> 3. 54.67 mg
}
2. 28.71 mg
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Set $y(t)$ equal to $0.9 \cdot y(0)$ and solve for $t$ :

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y(t)=0.1 \cdot y(0)=y(0) \cdot\left(\frac{1}{2}\right)^{\left(\frac{t}{t_{h}}\right)} \Rightarrow 0.1=e^{\ln \left(\frac{1}{2}\right)\left(\frac{t}{t_{h}}\right)}
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Taking the natrual logarithm of each side gives:

$$
\begin{gathered}
\ln 0.1=\ln \left(\frac{1}{2}\right)\left(\frac{t}{t_{h}}\right) \quad \text { so } \\
t=\frac{t_{h} \ln 0.1}{\ln \left(\frac{1}{2}\right)}=\frac{140 \cdot \ln 0.1}{\ln \left(\frac{1}{2}\right)}=465 \text { days }
\end{gathered}
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## Question 2

carbon- 14 has a half-life of 5730 years. How long does it take for $34 \%$ of a sample to decay?

1. $2042 y r s$
2. 6390yrs
3. 5531yrs
4. 7062yrs
5. 3435yrs
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4. 5531yrs
5. 3435yrs.

$$
t=\frac{5730 \cdot \ln (1-0.34)}{\ln \left(\frac{1}{2}\right)}=3435 \text { years }
$$

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This quantity is called the doubling time $t_{d}$ and is independent of the quantity present at the start:

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2 y(0)=y(0) e^{k t_{d}} \quad \text { so } \quad 2=e^{k t_{d}} \Rightarrow k=\frac{\ln 2}{t_{d}}
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Then the quantity present at time $t$ is:

$$
y(t)=y(0) e^{k t}=y(0) \cdot e^{\ln 2\left(\frac{t}{t_{d}}\right)}=y(0) \cdot(2)^{\left(\frac{t}{t_{d}}\right)}
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## Exponential Growth

A bacteria culture starts with 1000 cells and doubles every 2 hours. How many cells are there after 6 hours?

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The equation for the number of cells at time $t$ is:

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$$

The number of cells present after 6 hours is:

$$
y(6)=1000 \cdot(2)^{\left(\frac{6}{2}\right)}=8000 \text { cells }
$$

## Question 3

An experiment with fruit flies finds that the doubling time for the population of Bactrocera zonata on a wheat shorts diet is 2.78 days (Hussain, 1995). If we start with a population of 100 flies, what is the number of flies after 20 days?

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1. 14,645 <br> 2. 12,432 <br> 3. 13,167 <br> 4. 15,089 <br> 5. 16,006 <br> 6. cannot be determined
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1. 14,645 .

$$
y(20)=100 \cdot 2^{\frac{20}{2.78}}=14,645
$$

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Set $y(t)$ equal to 10,000 and solve for $t$ :

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y(t)=10000=200 \cdot(2)^{\left(\frac{t}{3}\right)} \Rightarrow 50=e^{\ln (2)\left(\frac{t}{3}\right)}
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$$

Taking the natrual logarithm of each side gives:

$$
\begin{aligned}
& \ln 50=\ln (2)\left(\frac{t}{3}\right) \text { so } \\
& t=\frac{3 \cdot \ln 50}{\ln 2}=16.93 \text { hours }
\end{aligned}
$$

## Question 4

An experiment with fruit flies finds that the doubling time for the population of Bactrocera zonata on a wheat shorts diet is 2.78 days (Hussain, 1995). If we start with a population of 100 flies, how many days does it take for the population to reach 4,000 flies?

```
1. 13.91
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2. 15.33
3. 14.79
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```


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$$
\begin{array}{llll}
\text { 1. } & 13.91 & \text { 4. } & 14.02 \\
\text { 2. } & 15.33 & \text { 5. } & 16.21 \\
\text { 3. } & 14.79 & \text { 6. } & \text { cannot be determined } \\
\text { 3. } & 14.79 . & \\
& t=\frac{2.78 \cdot \ln 40}{\ln 2}=14.79 \text { days }
\end{array}
$$

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Then

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y(t)=y(0) e^{k t}
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## Exponential Cooling

Suppose the heat is turned off under a pot of boiling water $\left(212^{\circ} \mathrm{F}\right)$ in a room with an ambient temperature of $70^{\circ} \mathrm{F}$. After 10 minutes, the temperature has dropped to $\left(160^{\circ} \mathrm{F}\right)$. What is the temperature of the water after 30 minutes?

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Let

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y=T-T_{\text {ambient }}
$$

Then

$$
y(0)=212-70=142
$$

and

$$
y(10)=160-70=90
$$

## Exponential Cooling

Our exponential cooling equation is then:

$$
y(t)=y(0) \cdot e^{k t} \quad \text { so } \quad 90=142 e^{k \cdot 10} \quad \text { and } \quad \frac{90}{142}=e^{k \cdot 10}
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Taking logs of both sides, we get

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\ln \frac{90}{142}=k \cdot 10 \quad \text { so } \quad k=\frac{\ln \frac{90}{142}}{10}=-0.0456
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Then

$$
y(t)=142 \cdot e^{-0.0456 t} \quad \text { and } \quad y(30)=142 \cdot e^{-0.0456 \cdot 30}=36.15
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Then

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y(t)=142 \cdot e^{-0.0456 t} \quad \text { and } \quad y(30)=142 \cdot e^{-0.0456 \cdot 30}=36.15
$$

So after 30 minutes, the water is $36.15^{\circ} \mathrm{F}$ above ambient, or $106.15^{\circ} \mathrm{F}$.

## Question 5

Cooling water from a power plant is pumped into a holding tank at $90^{\circ} \mathrm{C}$. After 20 minutes, the temperature has dropped to $81^{\circ} \mathrm{C}$. If the ambient temperature is $24^{\circ} \mathrm{C}$, what is the temperature of the water after 2 hours?

| 1. | 47.64 | 4. | 44.02 |
| :--- | :--- | :--- | :--- |
| 2. | 51.39 | 5. | 46.21 |
| 3. | 55.21 | 6. | cannot be determined |

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3. $\quad 55.21 \quad$ 6. cannot be determined
4. 51.39. $y(0)=90-24=66$ and $y(20)=81-24=57$. So

$$
\ln \frac{57}{66}=k \cdot 20 \quad \text { so } \quad k=\frac{\ln \frac{57}{66}}{20}=-0.00733
$$

then

$$
y(120)=66 \cdot e^{-0.00733 \cdot 120}=27.39 \quad T_{120}=27.39+24=51.39
$$

