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$$y(t) = y(0)e^{kt} = y(0) \cdot e^{\ln\left(\frac{1}{2}\right)\left(\frac{t}{t_h}\right)} = y(0) \cdot \left(\frac{1}{2}\right)^{\left(\frac{t}{t_h}\right)}$$

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The quantity remaining after 180 days is:

$$y(180) = 300 \cdot \left(\frac{1}{2}\right)^{\left(\frac{180}{140}\right)} = 123.05mg$$

bismuth-210 has a half-life of 5 days. How much of a sample of 200mg is left after 14 days?

- **1.** 16.48mg
- **2.** 92.57mg
- **3.** 54.67mg

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Set y(t) equal to $0.9 \cdot y(0)$ and solve for t:

$$y(t) = 0.1 \cdot y(0) = y(0) \cdot \left(\frac{1}{2}\right)^{\left(\frac{t}{t_h}\right)} \Rightarrow 0.1 = e^{\ln\left(\frac{1}{2}\right)\left(\frac{t}{t_h}\right)}$$

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Taking the natrual logarithm of each side gives:

$$\ln 0.1 = \ln \left(\frac{1}{2}\right) \left(\frac{t}{t_h}\right) \quad \text{so}$$

$$t = \frac{t_h \ln 0.1}{\ln \left(\frac{1}{2}\right)} = \frac{140 \cdot \ln 0.1}{\ln \left(\frac{1}{2}\right)} = 465 \text{ days}$$

carbon-14 has a half-life of 5730 years. How long does it take for 34% of a sample to decay?

- **1.** 2042*yrs* **4.** 7062*yrs*
- **2.** 6390*yrs*

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- **5.** 3435*yrs*
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$$t = \frac{5730 \cdot \ln(1 - 0.34)}{\ln\left(\frac{1}{2}\right)} = 3435 \text{ years}$$

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 so $2 = e^{kt_d} \Rightarrow k = \frac{\ln 2}{t_d}$

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Then the quantity present at time t is:

$$y(t) = y(0)e^{kt} = y(0) \cdot e^{\ln 2\left(\frac{t}{t_d}\right)} = y(0) \cdot (2)^{\left(\frac{t}{t_d}\right)}$$

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The equation for the number of cells at time t is:

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The number of cells present after 6 hours is:

$$y(6) = 1000 \cdot (2)^{\left(\frac{6}{2}\right)} = 8000$$
 cells

An experiment with fruit flies finds that the doubling time for the population of Bactrocera zonata on a wheat shorts diet is 2.78 days (Hussain, 1995). If we start with a population of 100 flies, what is the number of flies after 20 days?

- **1.** 14,645 **4.** 15,089
- **2.** 12, 432 **5.** 16,006
- **3.** 13, 167 **6.** cannot be determined

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13,167

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$$y(20) = 100 \cdot 2^{\frac{20}{2.78}} = 14,645$$

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Taking the natrual logarithm of each side gives:

$$\ln 50 = \ln \left(2\right) \left(\frac{t}{3}\right) \quad \text{so}$$

$$t = \frac{3 \cdot \ln 50}{\ln 2} = 16.93$$
 hours

An experiment with fruit flies finds that the doubling time for the population of Bactrocera zonata on a wheat shorts diet is 2.78 days (Hussain, 1995). If we start with a population of 100 flies, how many days does it take for the population to reach 4,000 flies?

- **1.** 13.91 **4.** 14.02
- **2.** 15.33 **5.** 16.21
- **3.** 14.79 **6.** cannot be determined

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3. 14.79.

$$t = \frac{2.78 \cdot \ln 40}{\ln 2} = 14.79 \text{ days}$$

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$$y = T - T_{ambient}$$

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Then

$$y(t) = y(0)e^{kt}$$

Suppose the heat is turned off under a pot of boiling water $(212^{\circ}F)$ in a room with an ambient temperature of $70^{\circ}F$. After 10 minutes, the temperature has dropped to $(160^{\circ}F)$. What is the temperature of the water after 30 minutes?

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Let

$$y = T - T_{ambient}$$

Then

$$y(0) = 212 - 70 = 142$$

and

$$y(10) = 160 - 70 = 90$$

Our exponential cooling equation is then:

$$y(t) = y(0) \cdot e^{kt}$$
 so $90 = 142e^{k \cdot 10}$ and $\frac{90}{142} = e^{k \cdot 10}$

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Taking logs of both sides, we get

$$\ln\frac{90}{142} = k \cdot 10 \quad \text{so} \quad k = \frac{\ln\frac{90}{142}}{10} = -0.0456$$

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Then

 $y(t) = 142 \cdot e^{-0.0456t}$ and $y(30) = 142 \cdot e^{-0.0456 \cdot 30} = 36.15$

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So after 30 minutes, the water is $36.15^{\circ}F$ above ambient, or $106.15^{\circ}F$.

Cooling water from a power plant is pumped into a holding tank at $90^{\circ}C$. After 20 minutes, the temperature has dropped to $81^{\circ}C$. If the ambient temperature is $24^{\circ}C$, what is the temperature of the water after 2 hours?

- **1.** 47.64 **4.** 44.02
- **2.** 51.39 **5.** 46.21
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- **2.** 51.39 **5.** 46.21
- **3.** 55.21 **6.** cannot be determined

2. 51.39. y(0) = 90 - 24 = 66 and y(20) = 81 - 24 = 57. So

$$\ln \frac{57}{66} = k \cdot 20 \quad \text{so} \quad k = \frac{\ln \frac{57}{66}}{20} = -0.00733$$

then

$$y(120) = 66 \cdot e^{-0.00733 \cdot 120} = 27.39$$
 $T_{120} = 27.39 + 24 = 51.39$